



From Failure to Success: Orthogonal Area-Selective Thin Film Deposition

Gregory N. Parsons

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Landon Keller, Hannah Margavio, Holger Saare

*Department of Chemical and Biomolecular Engineering
North Carolina State University, Raleigh, NC*

SRC Tasks: 2729.001, 2873.001, 3036.001

SRC Seminar

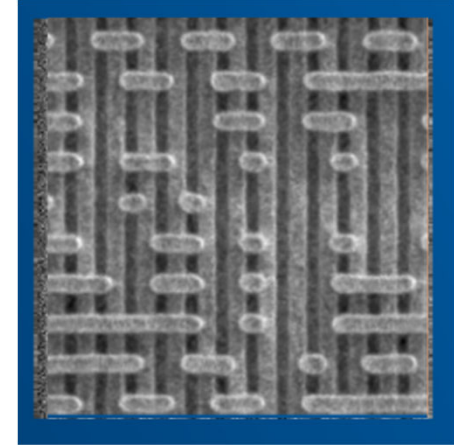
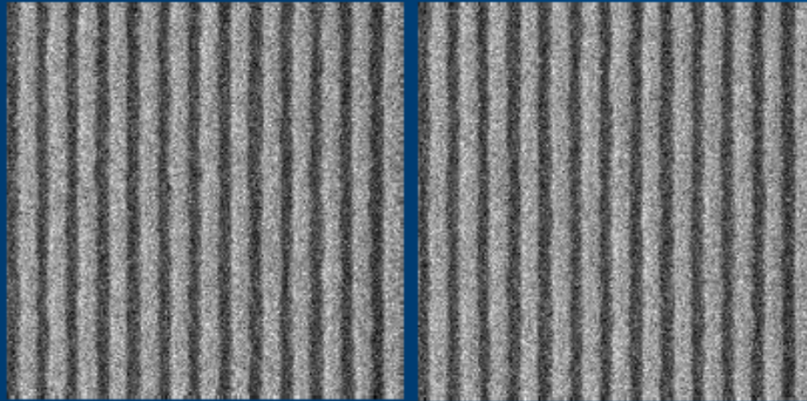
Sept 23, 2021

Outline

- Motivation: Deposition-based patterning
- Idea: Orthogonal ASD - Two ASD processes “back-to-back”
- Why we thought our idea would work
 - Success criteria
- Why our original idea was not successful
 - Would the direction have been successful with different boundary conditions?
- How we worked around the problem
 - Discovered a new direction
- What we learned...
 - Success criteria



dipole
illumination
26 nm pitch

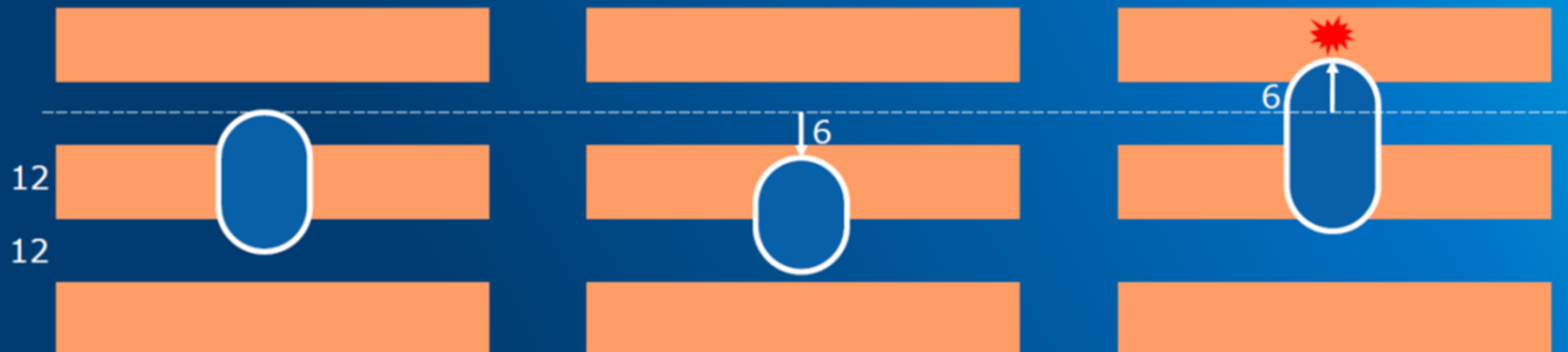


EPE control is critical

Nominal Cut

Incomplete Cut = SHORT

Unintended Cut = Fill / Rel. Problem

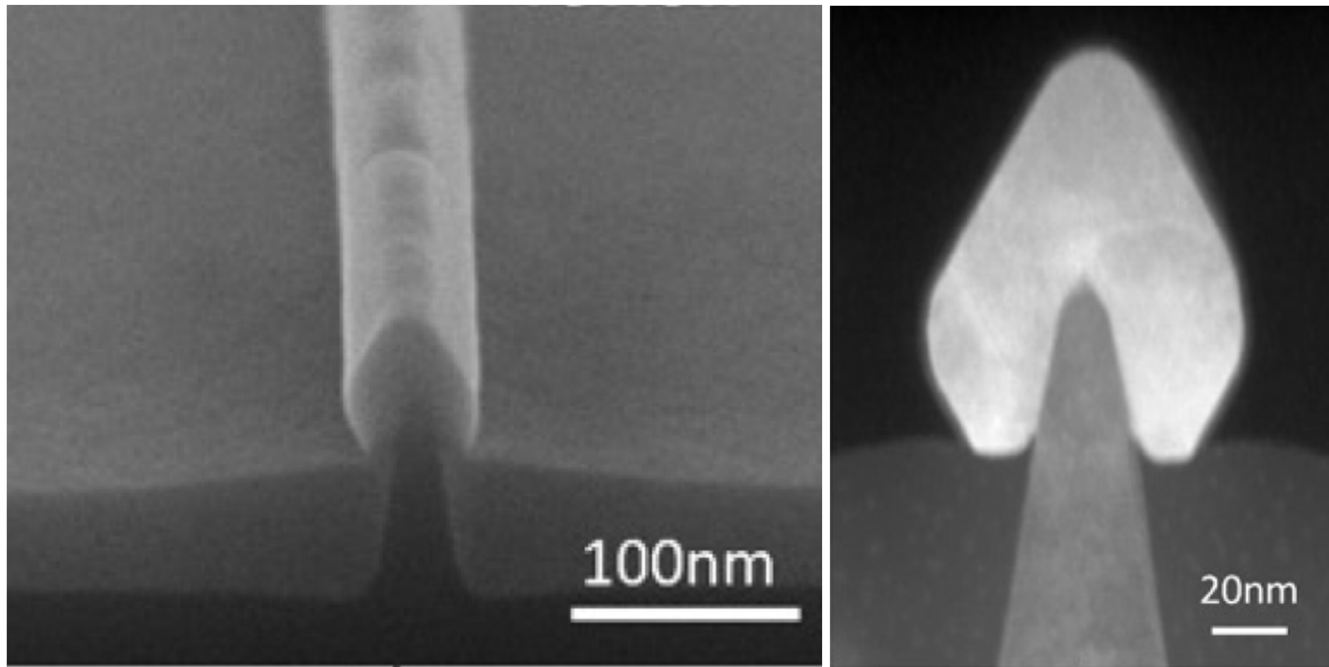


Charles Wallace

SPIE Advanced Lithography, EUV, Feb. 24, 2020

Deposition-Based Patterning

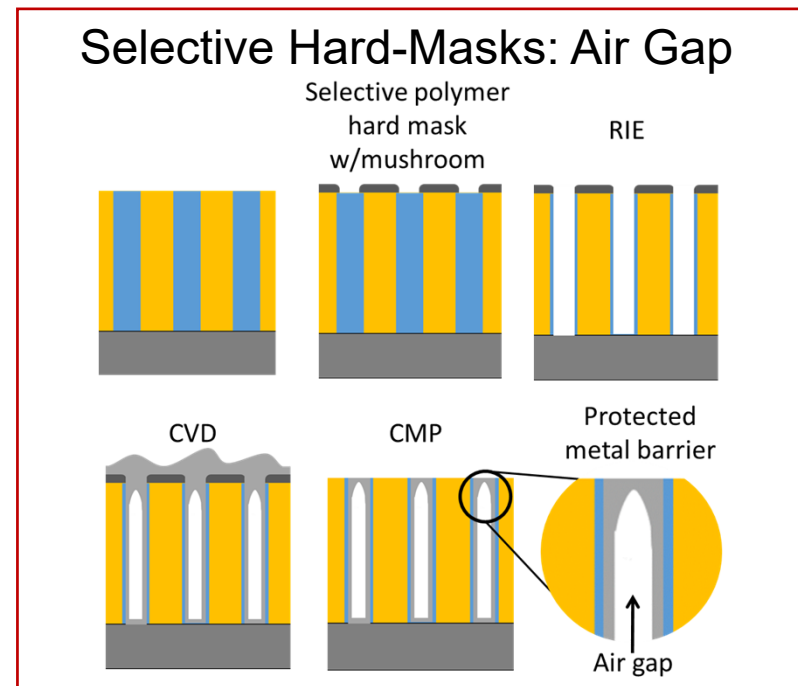
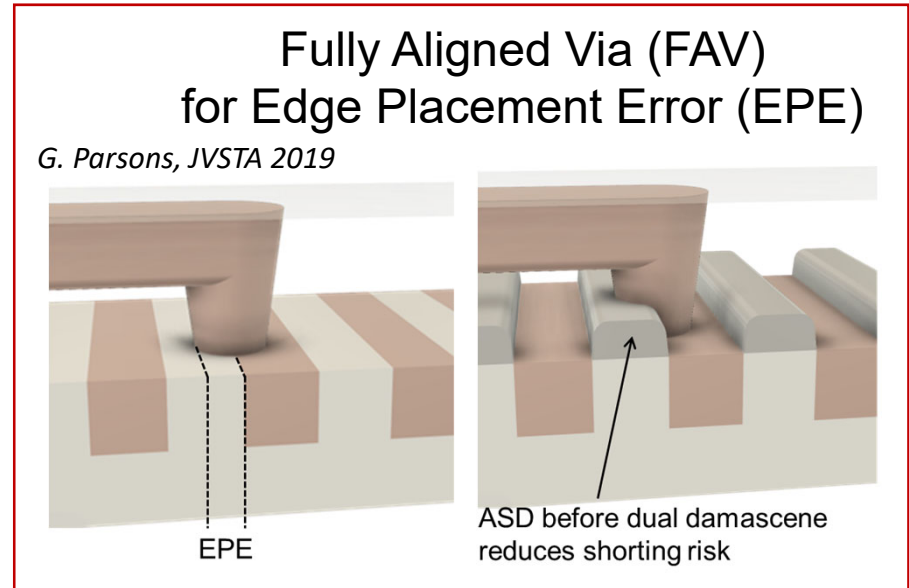
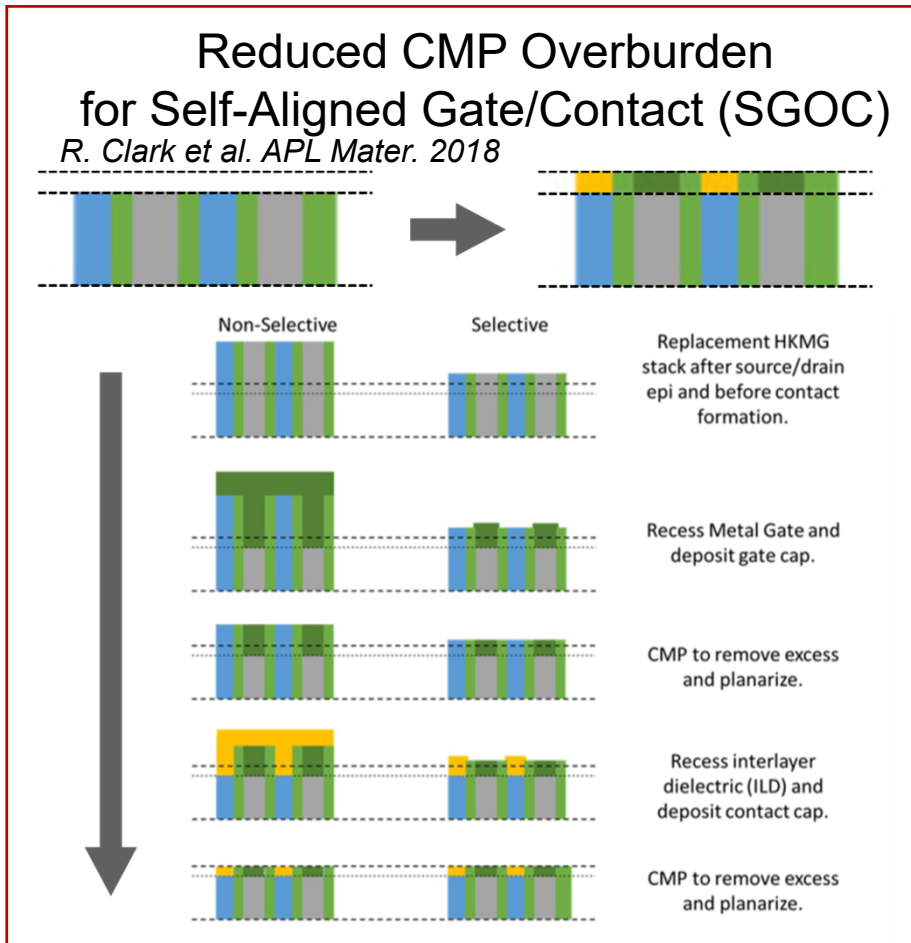
Selective Si, SiGe Epitaxy



G. Wang et al., *Microelectron. Eng.* (2016)

- Temperature = 700-800°C
- Thermodynamic control of surface termination
- For Back-end ASD: Lower temperatures requires control of reaction kinetics

Back-End Applications for Area Selective Deposition

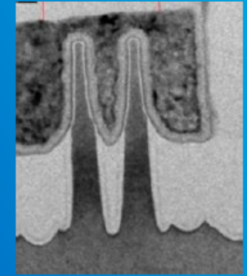


- Self-Aligned Block
- Back-end super-via
- Buried power
- Resist augmentation...other....



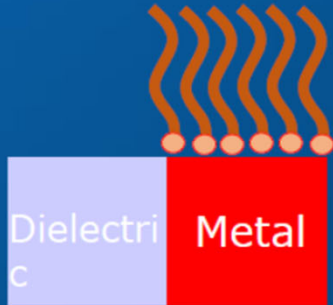
Self-Aligned Contact/COAG Challenges

Contact over active gate

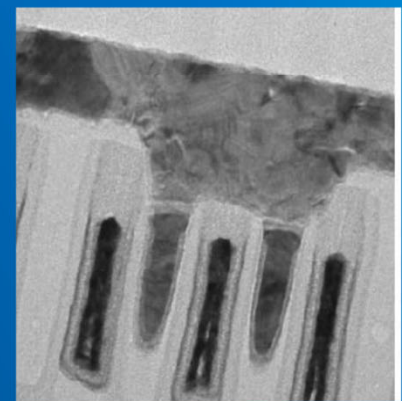
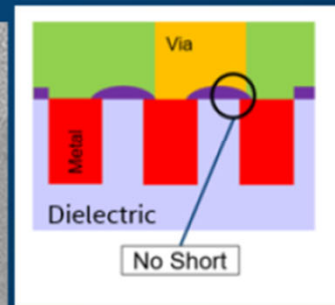
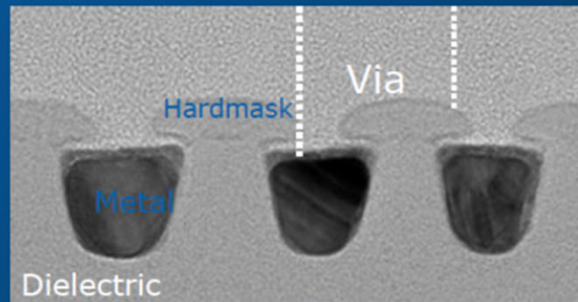


- Coloring is required when EPE control does not scale with pitch
- Coloring is expensive, complex. Many processing steps.

SAM Passivation

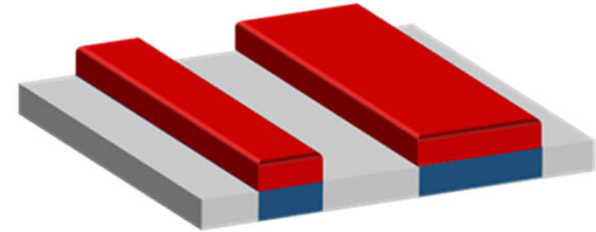
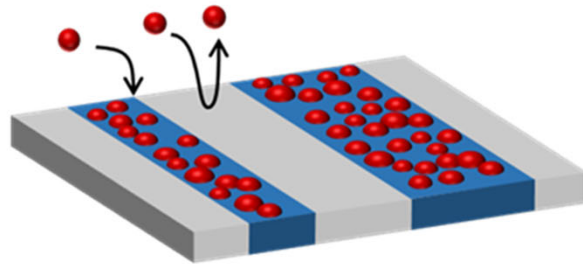


Metal Passivation

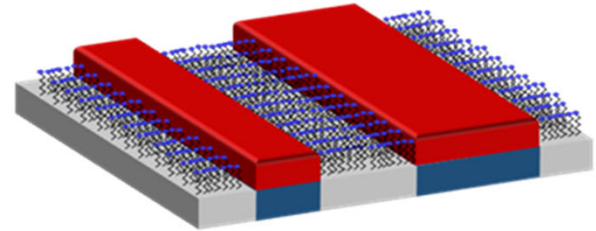
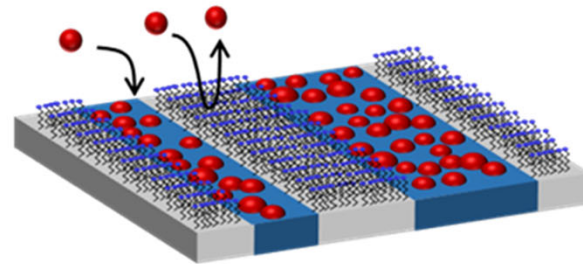


Approaches for Area-Selective Deposition

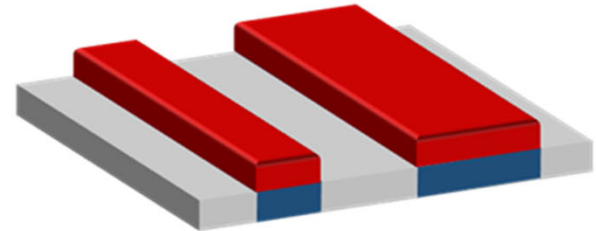
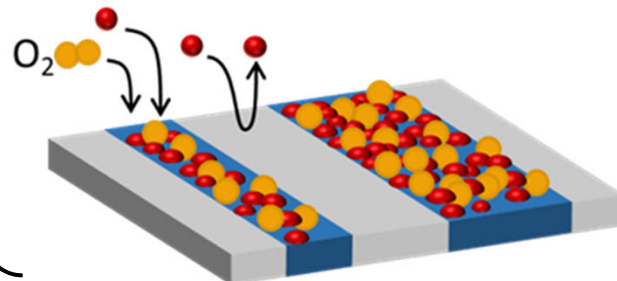
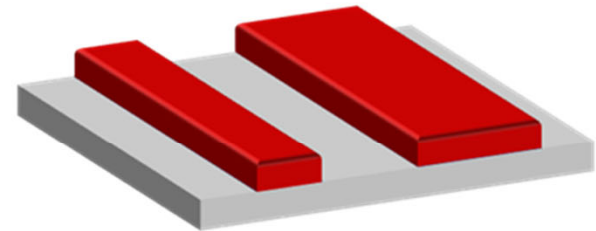
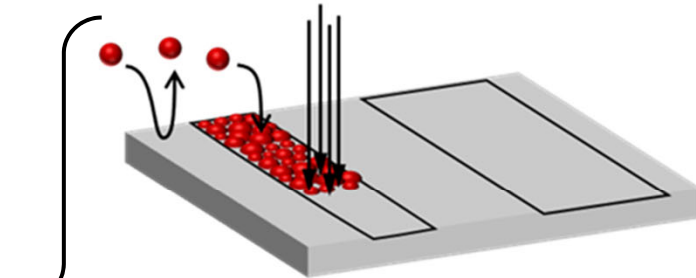
Intrinsic
(clean)



Passivated

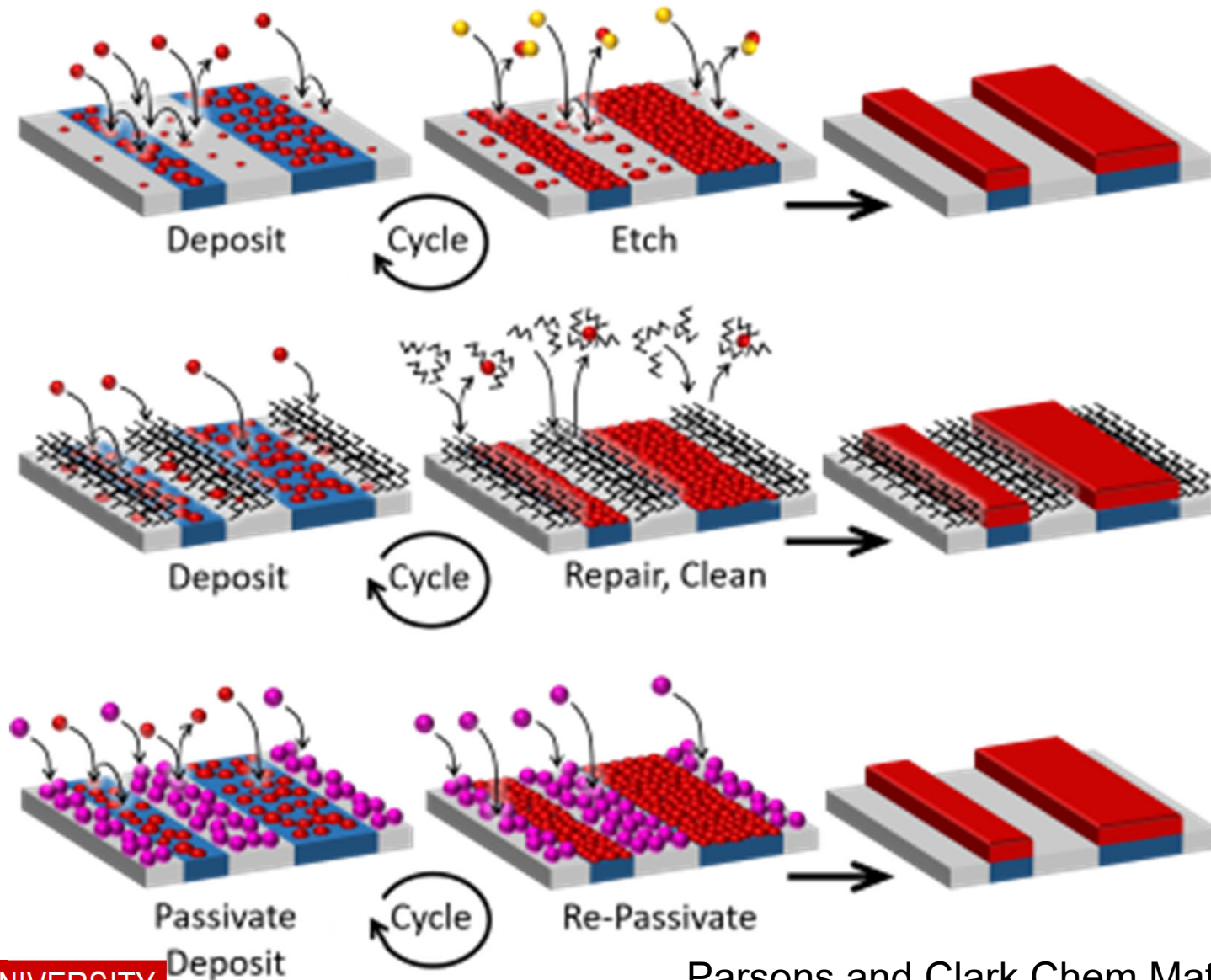


Activated

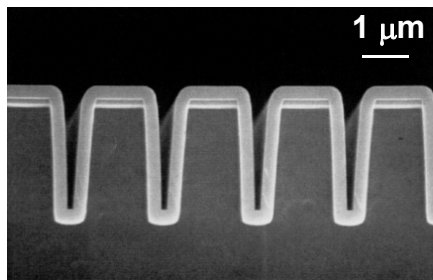
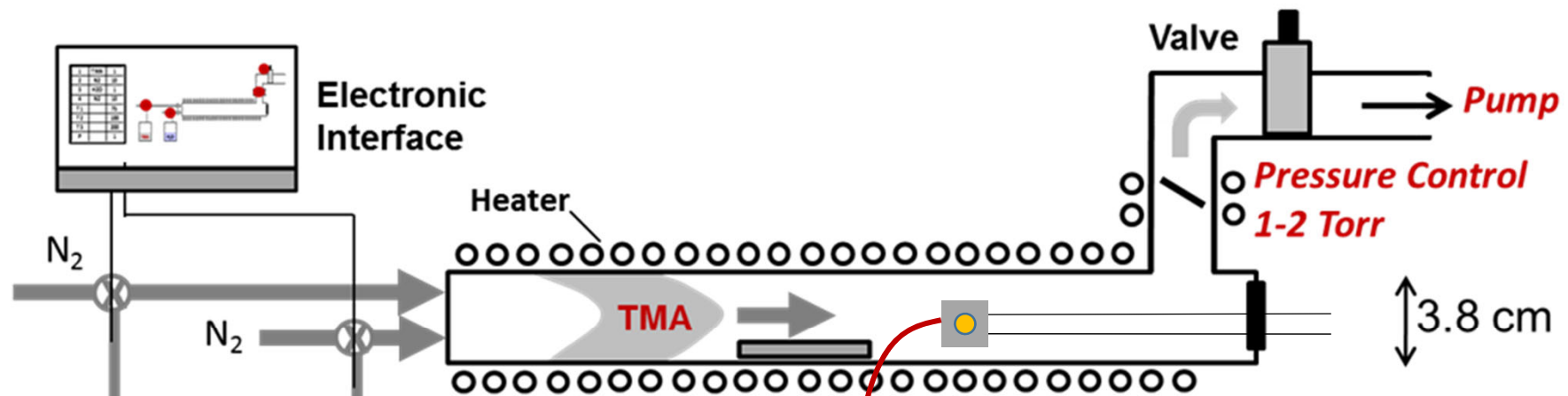


Approaches for Area-Selective Deposition

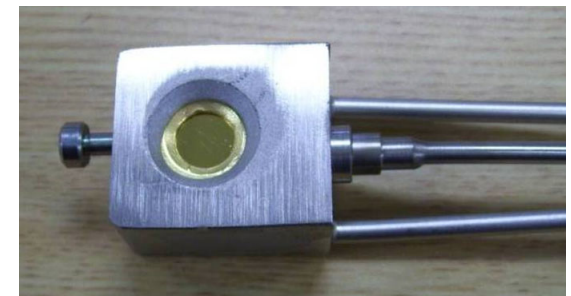
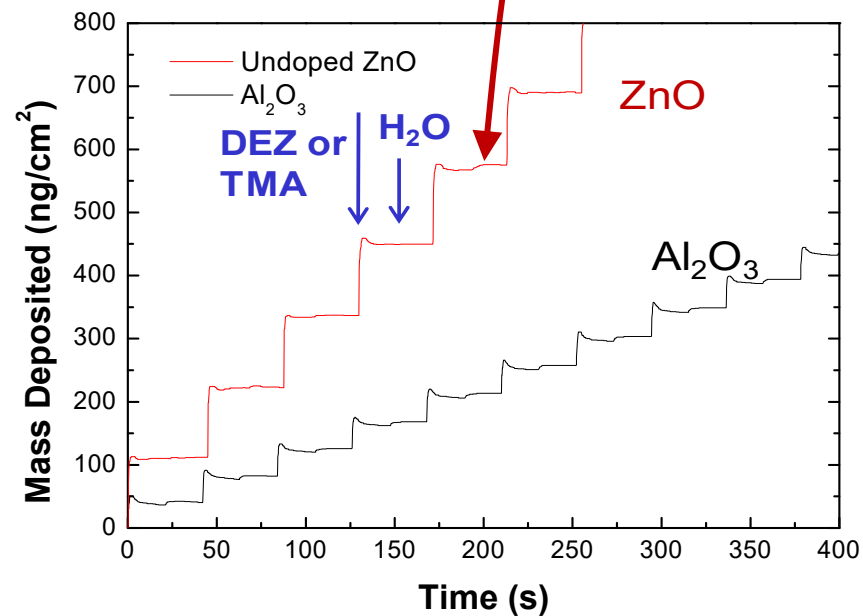
Multi-Step “Super-Cycle” Methods



Parsons Group: Lab-built ALD/ALE Reactors



Ritala, M. et al., *Chem. Vap. Dep.* 5 7 (1999)



Quartz Crystal Microbalance



W ASD on Si-H vs SiO₂

OVERVIEW OF RESEARCH

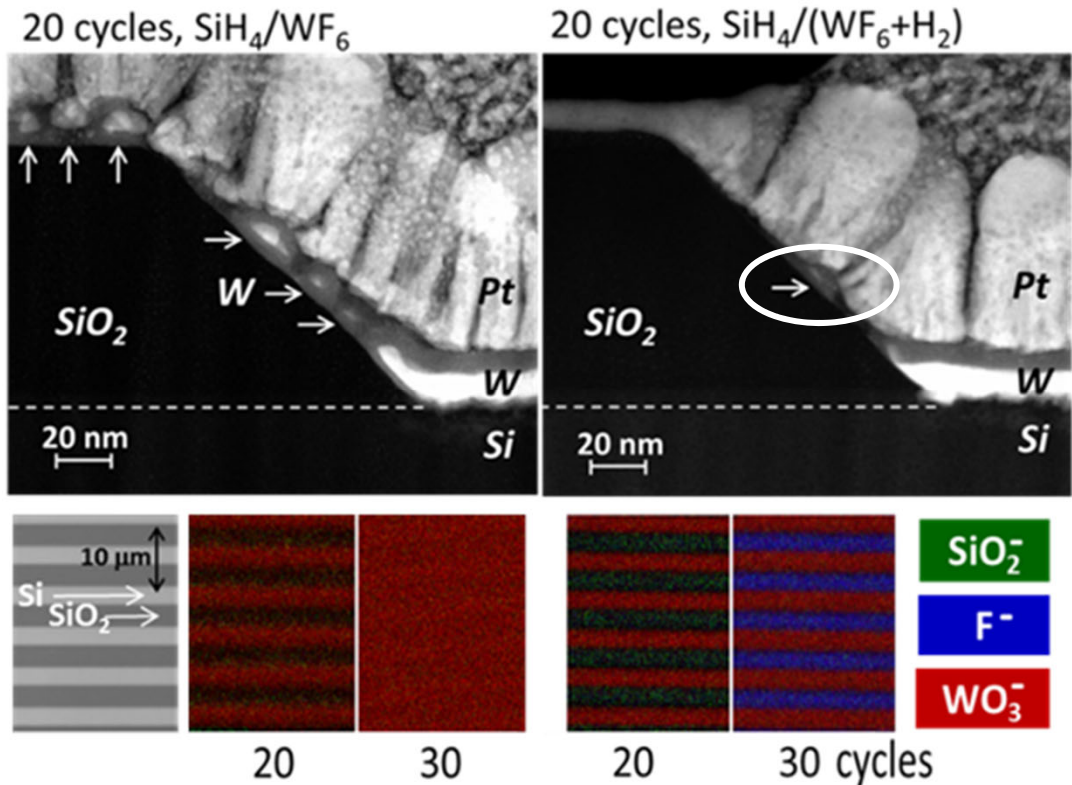
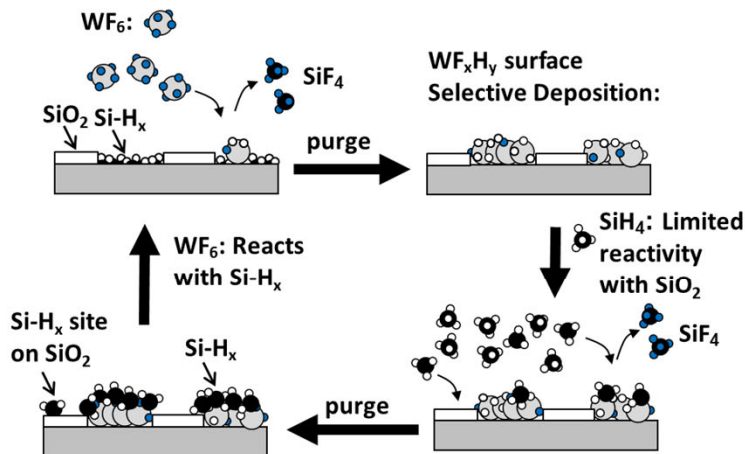
Project title: Selective Atomic Layer Deposition for Etch-Free Patterning

Investigators: Gregory N. Parsons

University: North Carolina State University

Topic: 2.4 Directed self-assembly of nanomaterials (etch-free patterning)

Date: Nov. 12, 2012



(19) **United States**

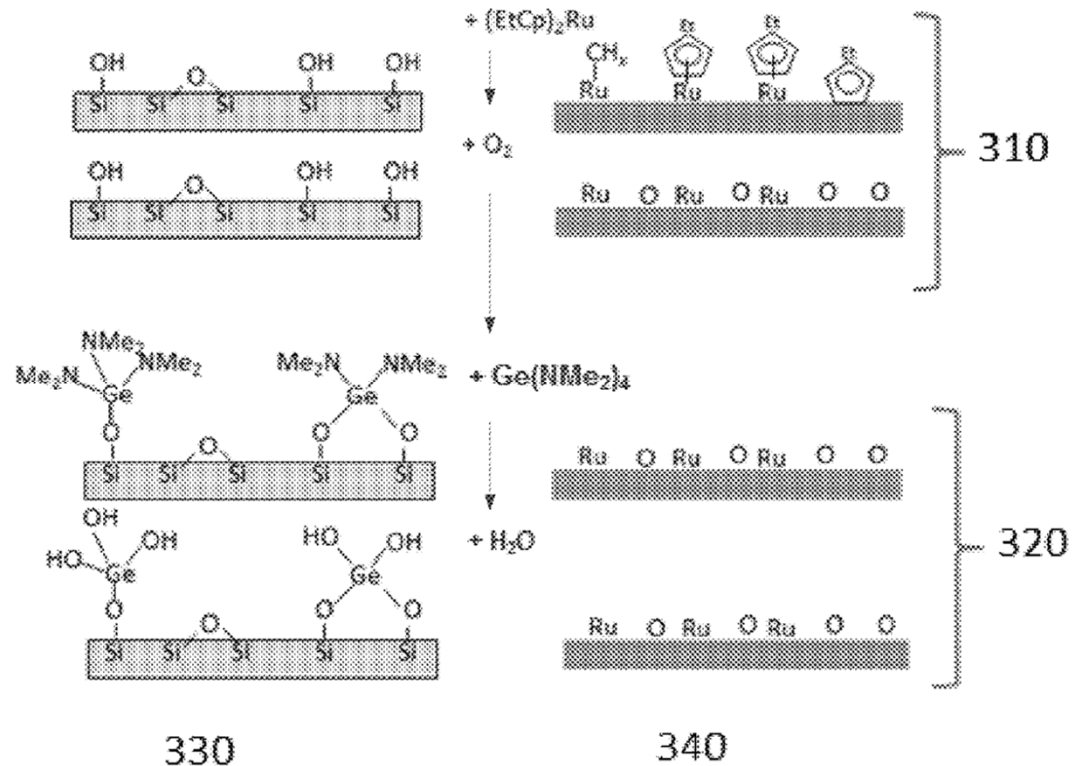
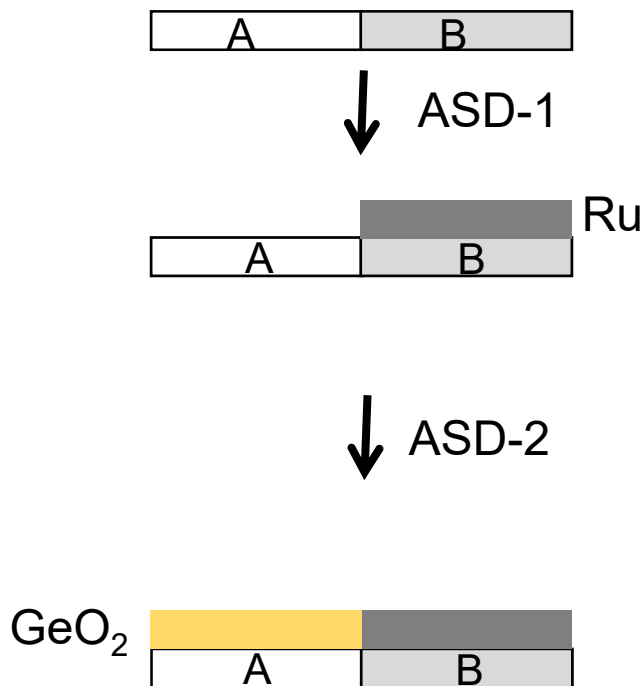
(12) **Patent Application Publication**
Haukka et al.

(10) **Pub. No.: US 2015/0299848 A1**

(43) **Pub. Date: Oct. 22, 2015**

(54) **DUAL SELECTIVE DEPOSITION**

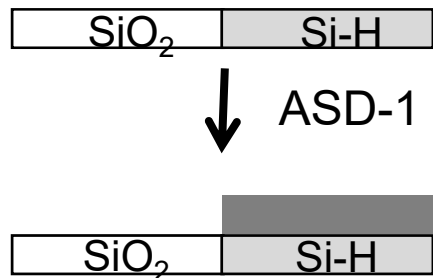
No data included in patent!



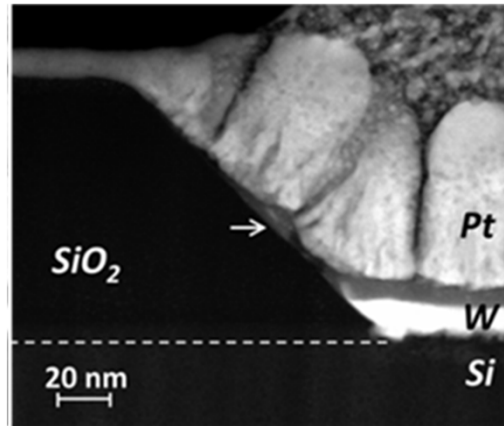
Idea

OVERVIEW OF RESEARCH

Project title: Orthogonal Dielectric/Metal Selective Area Atomic Layer Deposition
Investigators: Gregory N. Parsons
University: North Carolina State University
Topic: NMP #1: Selective deposition, growth, and removal as an enabler of self-aligned patterning strategies
Date: August 26, 2015

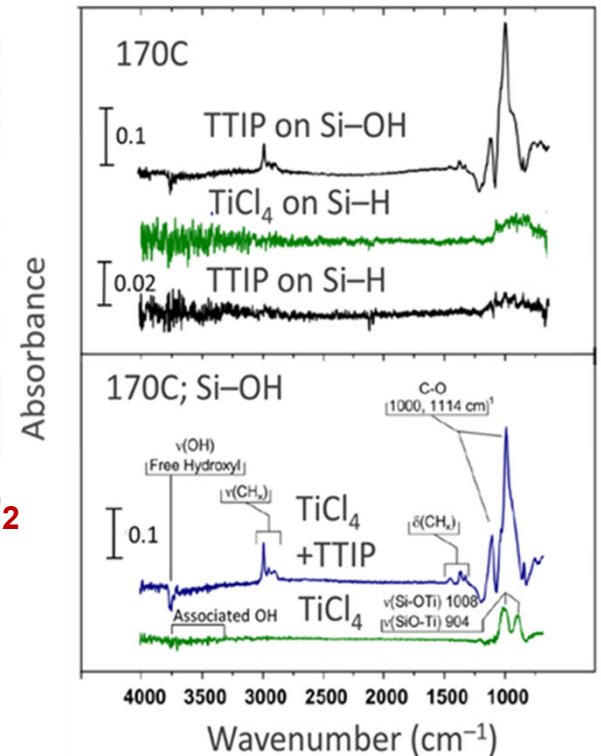


W?



W ASD on **Si-H vs SiO₂**
 B. Kalanyan et al.
 Chem Mat (2016)

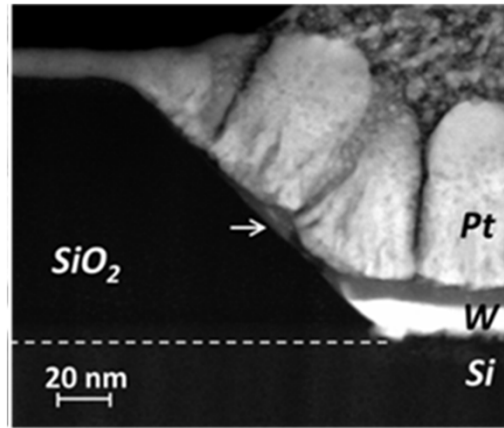
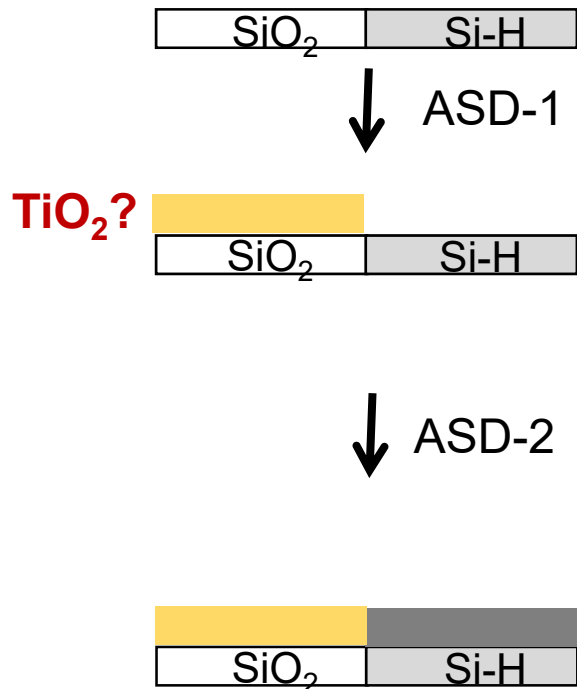
*“Metal-first”
 approach*



TiO₂ ASD on **SiO₂ vs Si-H**
 S. Atanasov et al.
 JVSTA (2016)

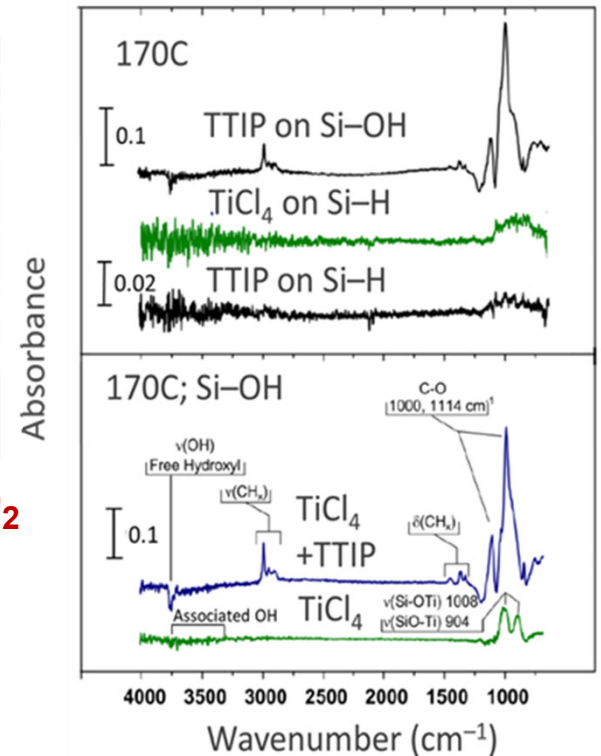
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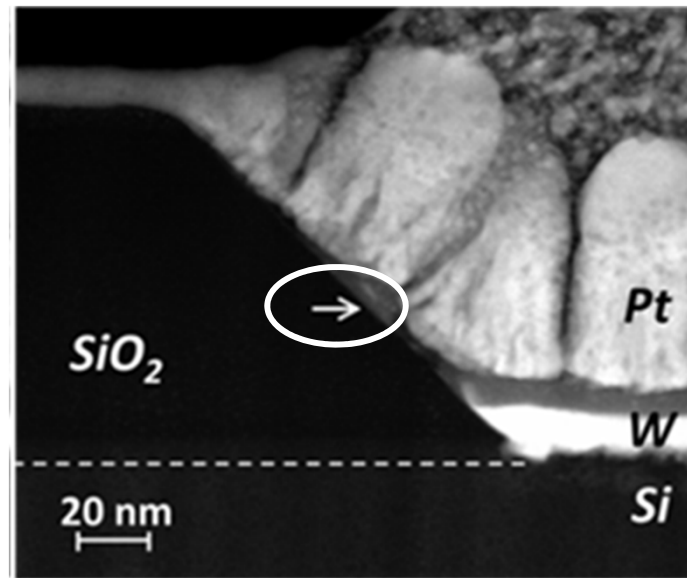
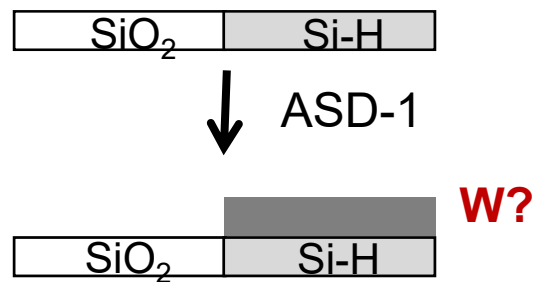


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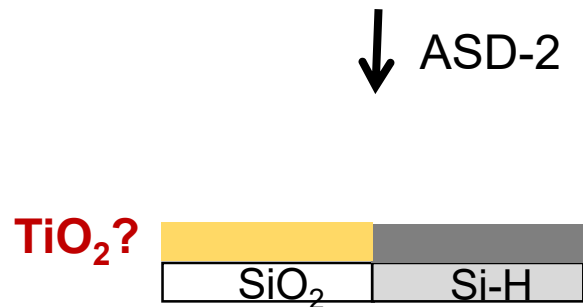
Success Criteria

1. Create new knowledge about surface reaction processes important for ASD
2. Achieve orthogonal ASD of two materials side-by-side (W and TiO_2) with thickness of 5-10 nm each
3. Build up multiple orthogonal layers, ideally 3 or more layers of each ASD material.

“Metal-First” Approach



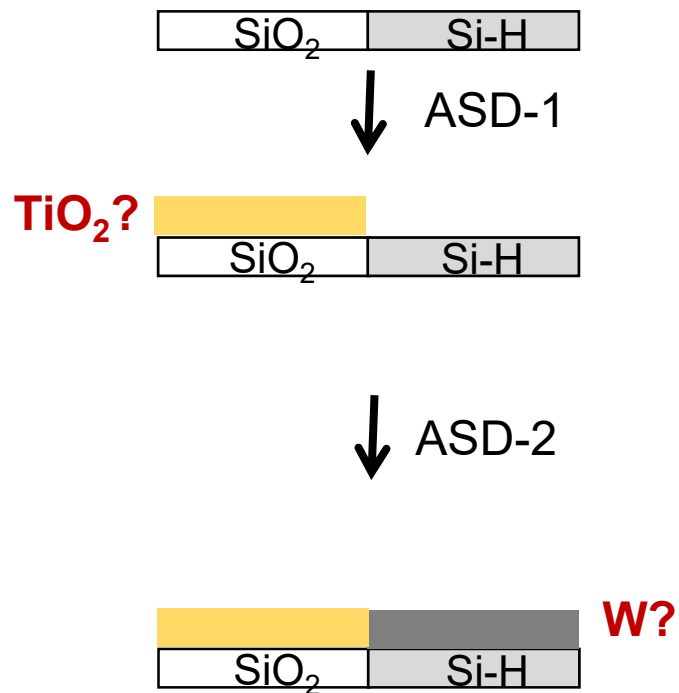
W ASD on **Si-H vs SiO₂**



TiO₂ grows readily on W

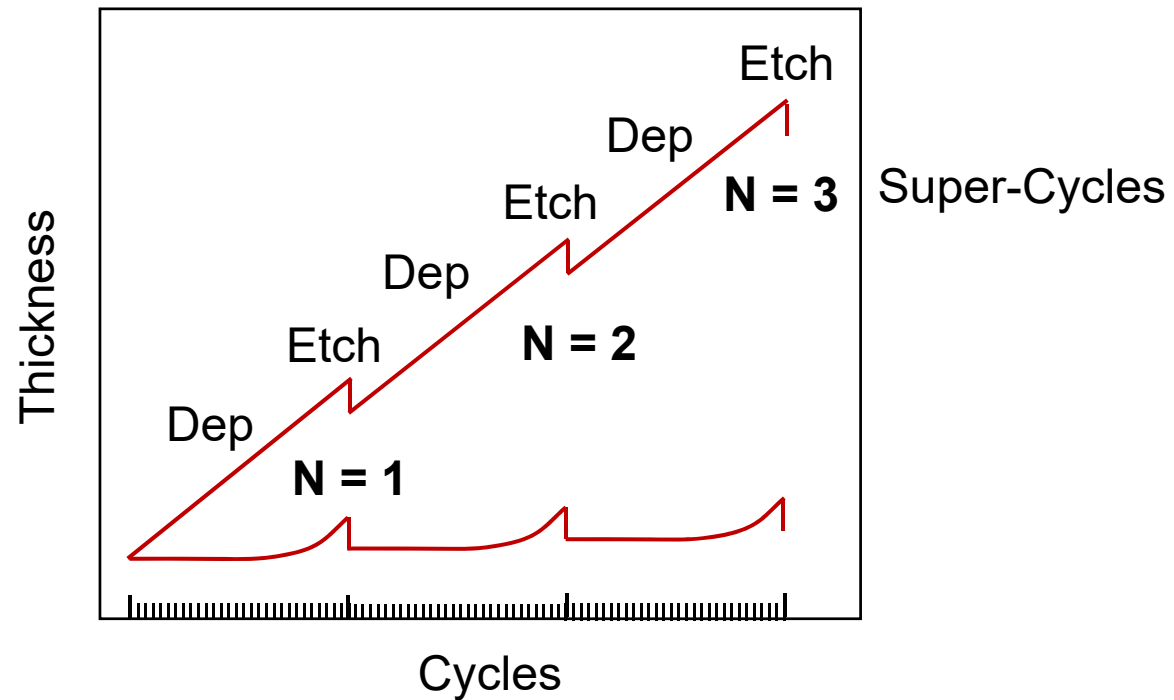
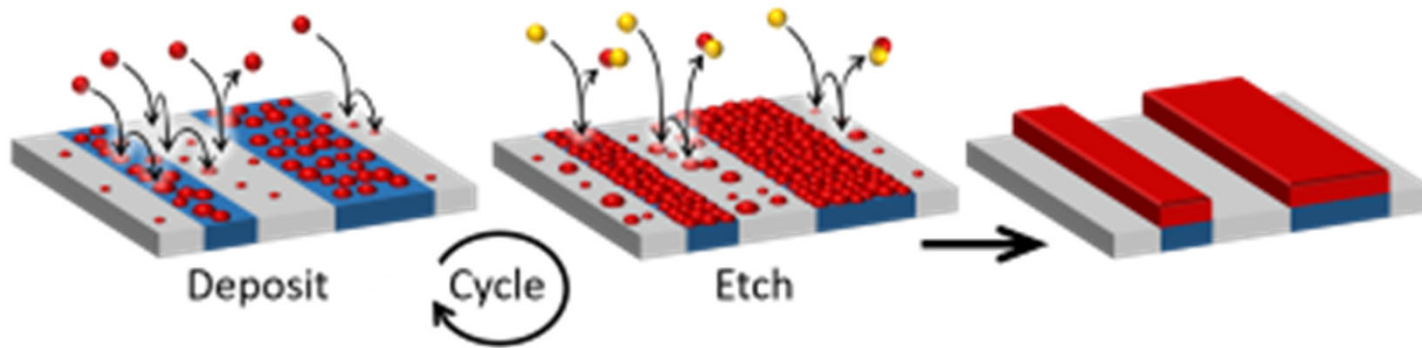
Failure

“Dielectric-First” Approach



Approaches for Area-Selective Deposition

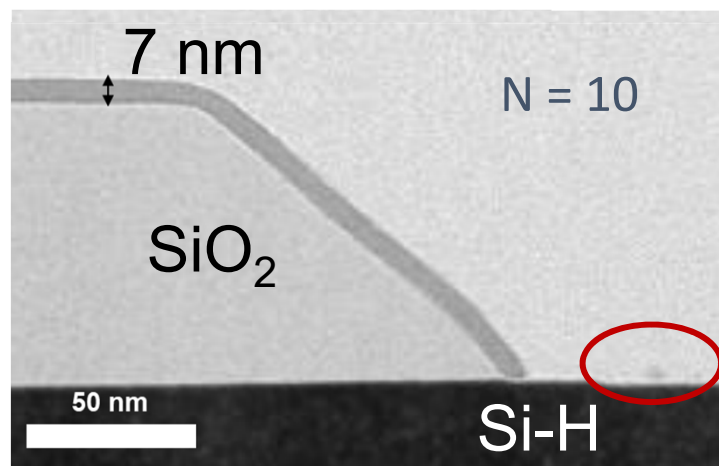
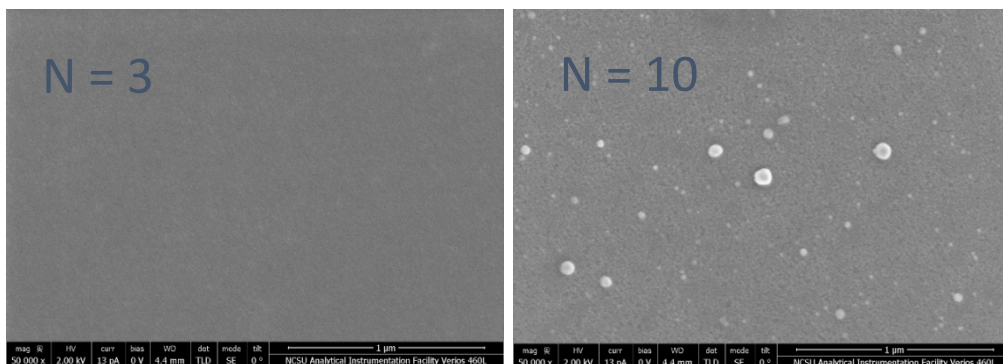
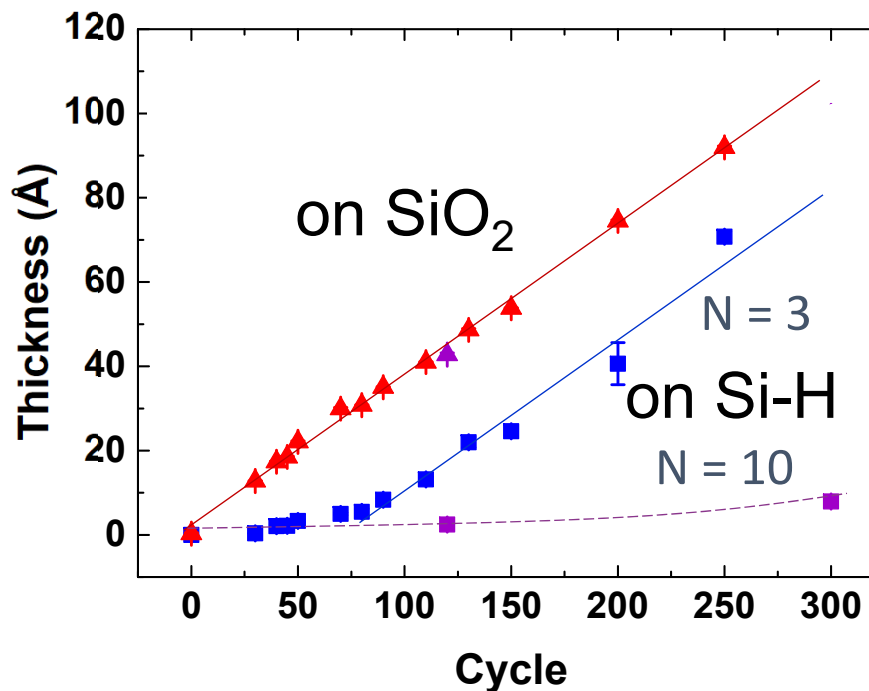
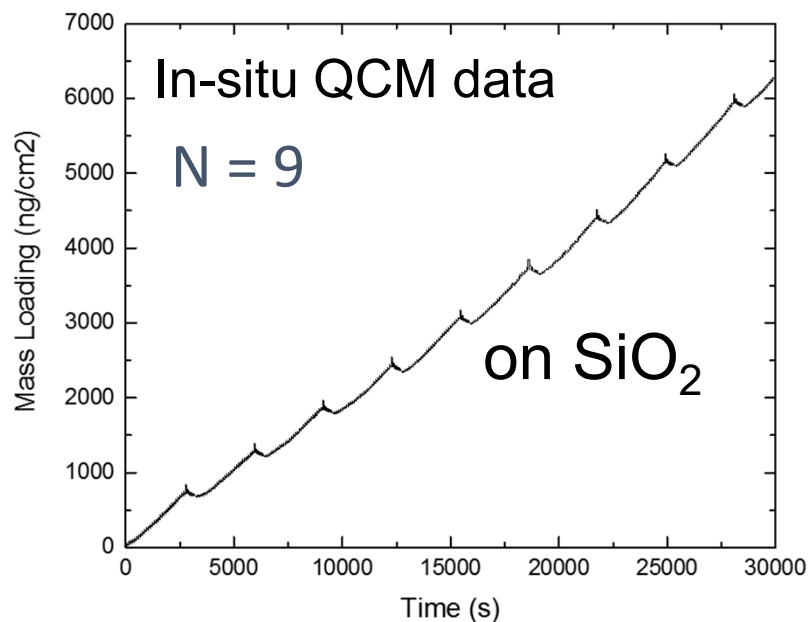
Multi-Step "Super-Cycle" Methods



TiO₂ ALD + ALE → ASD on SiO₂ vs Si-H

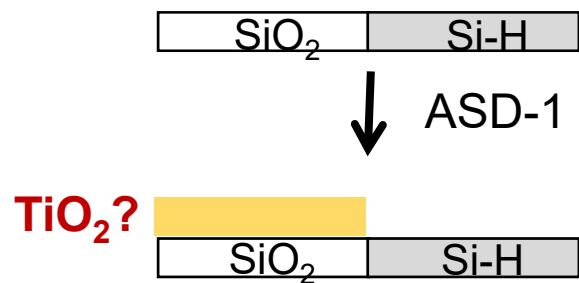
[30x(TiCl₄/H₂O) + 5x(WF₆/BCl₃)] × N = 10

2017

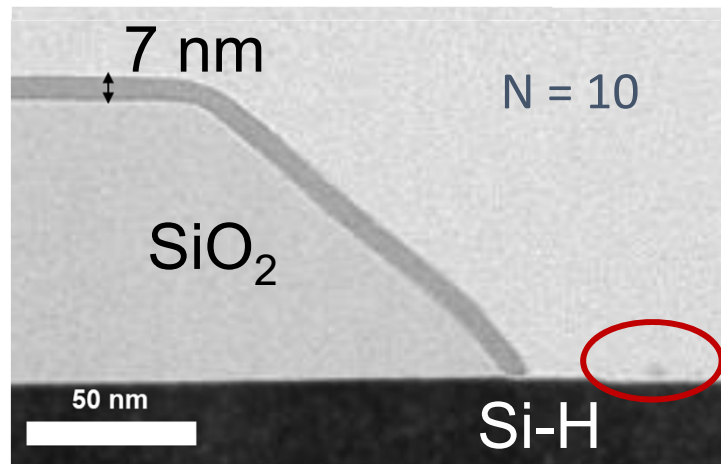
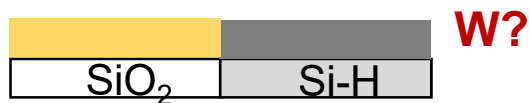


SK Song et al. Chem Mat (2019)

“Dielectric-First” Approach



ASD-2

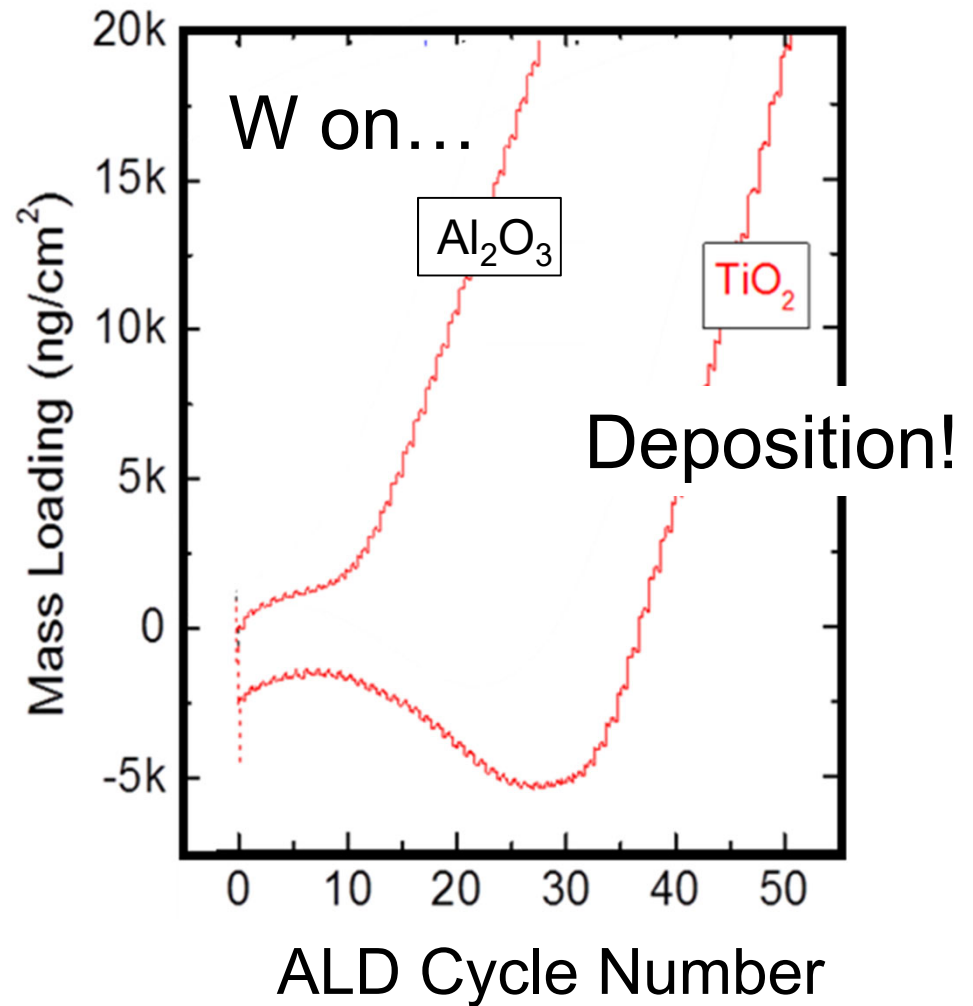
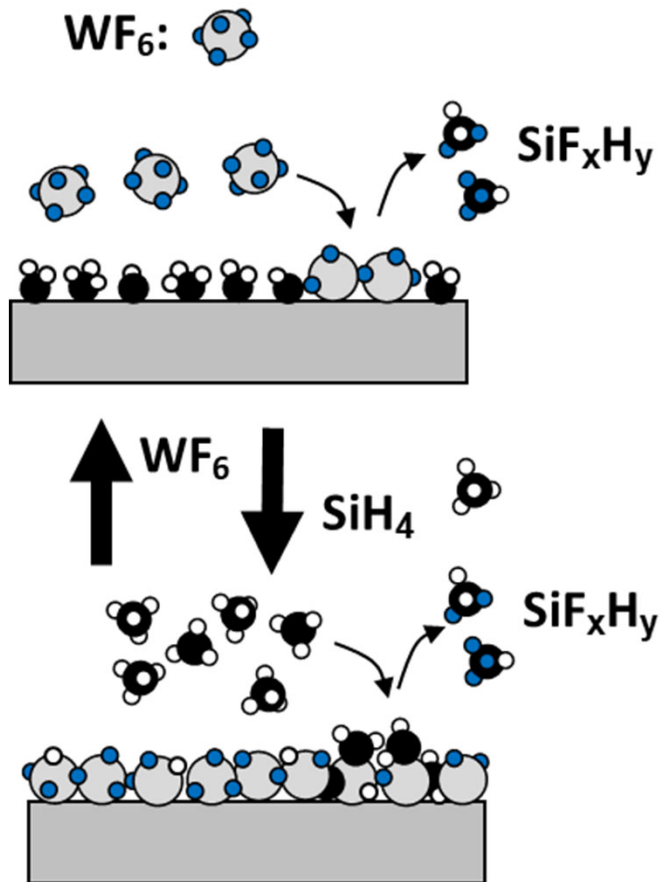


~ 7 nm

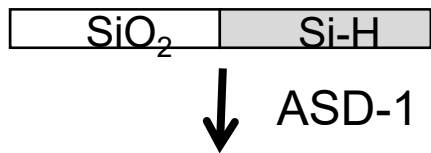
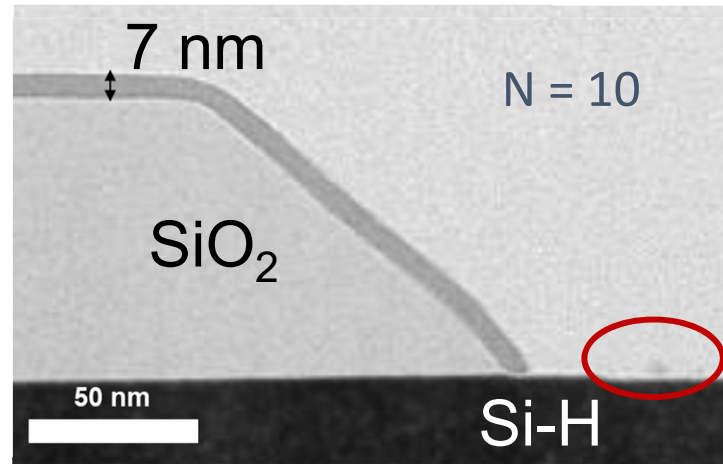
TiO₂ ASD on **SiO₂ vs Si-H**

W growth on
TiO₂?

W ASD on Si-H vs TiO₂?



“Dielectric-First” Approach



ASD-1

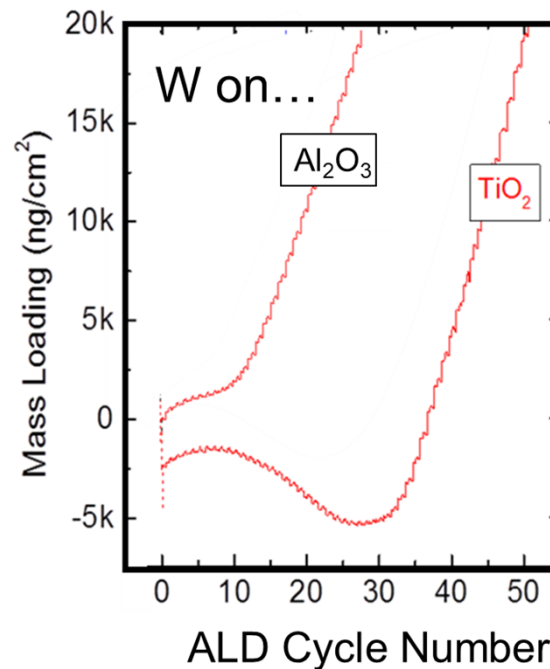
TiO₂?



ASD-2



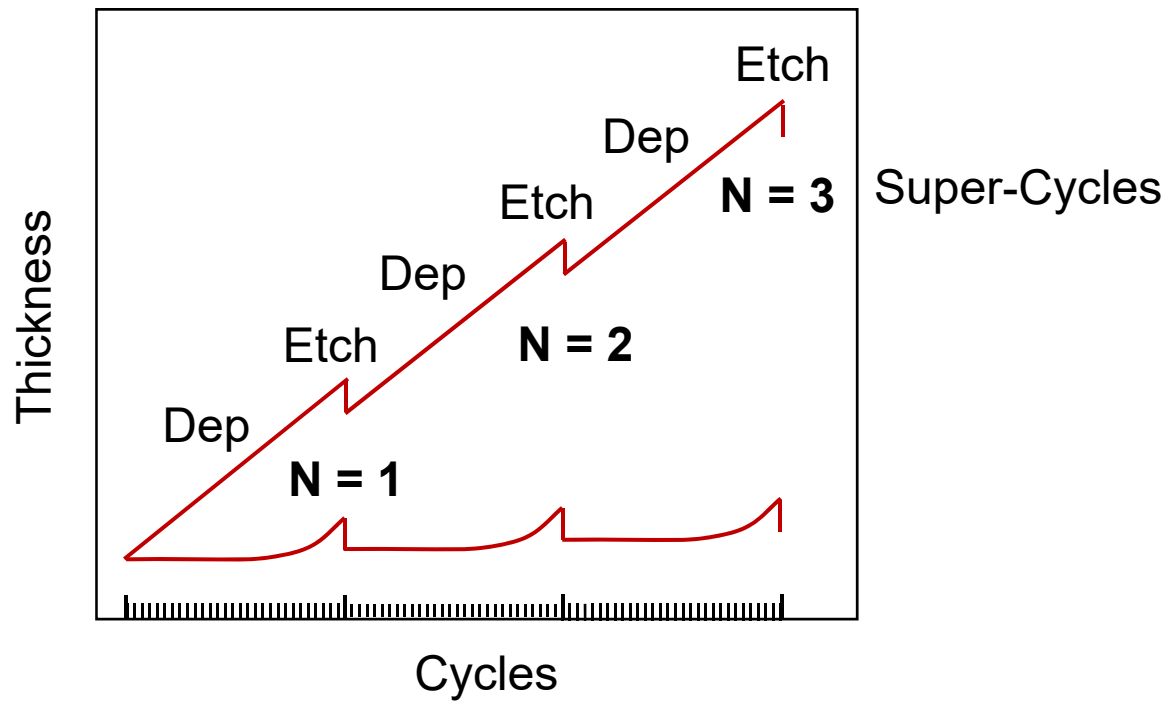
W?



W grows on TiO₂
Failure

What about W ASD via ALD + ALE?

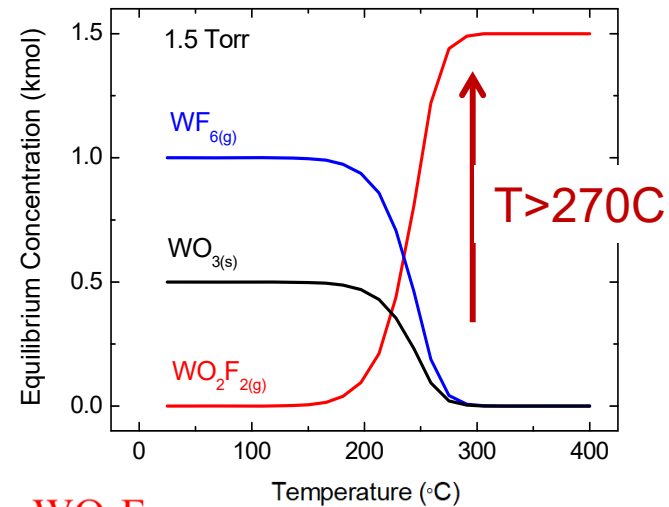
- If we can do W ASD using ALD+ALE, then maybe dielectric first approach could work



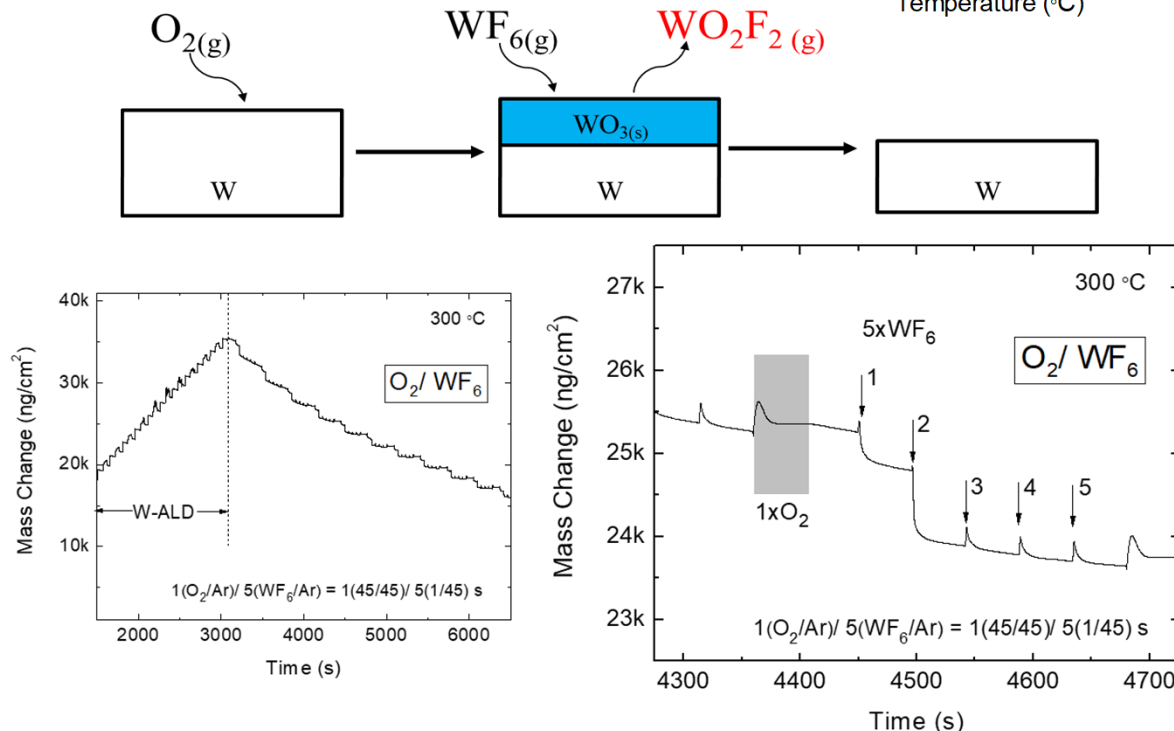
- Need ALE method for W

Thermodynamics of Tungsten ALE

- **Thermochemistry modeling**
 - $\text{WO}_3(\text{s})$: oxidized W product
 - $\text{WO}_2\text{F}_2(\text{g})$: etch product
 - $T \geq 150 \text{ }^\circ\text{C}$: WO_2F_2 formed
 - $T \rightarrow 300 \text{ }^\circ\text{C}$: WO_3 completely reacted
 - Also works with ozone

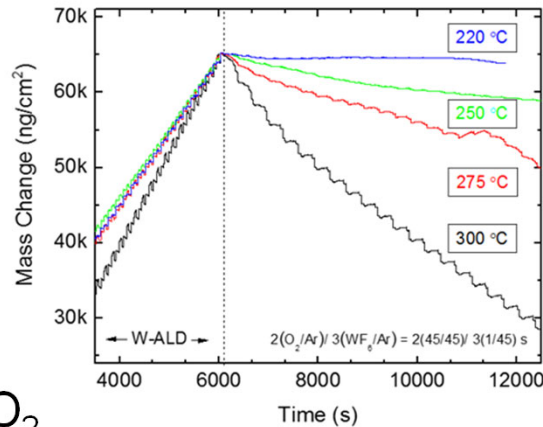


- **Two-step thermal W ALE**



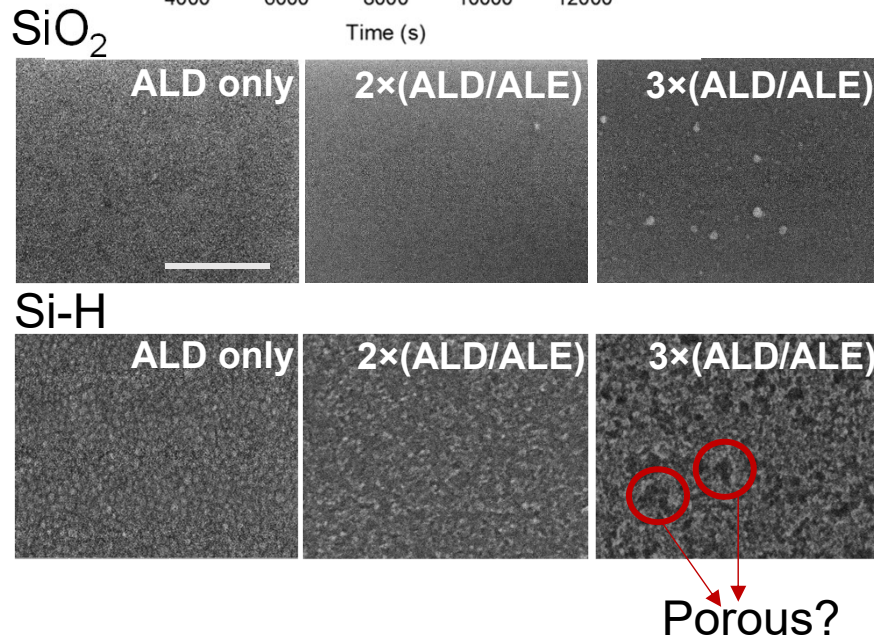
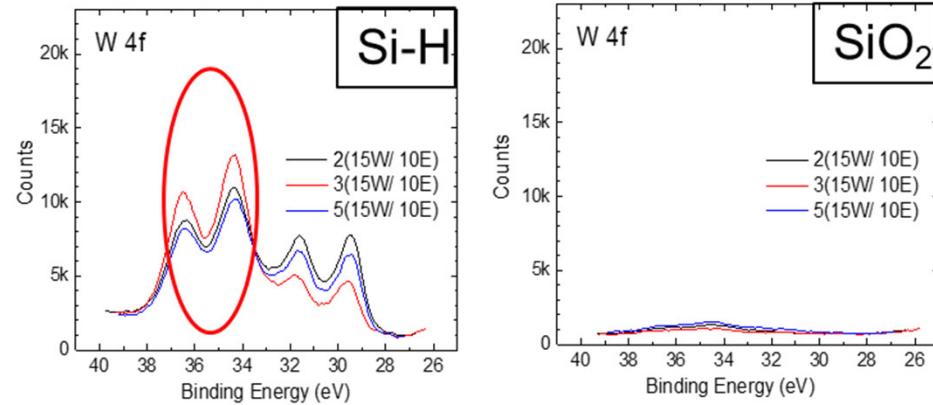
W: ALD + ALE = Poor ASD

Good W ALE: O_2/WF_6 at 275C
2018
 Wenyi Xie

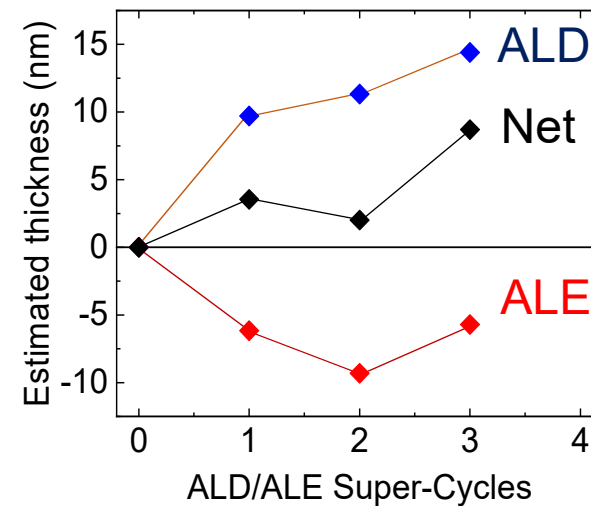


Wenyi Xie & Justin Kim 2019
 Not published

ALD+ALE Super-cycles:
 W growth slows on Si-H, ALE slows on SiO₂

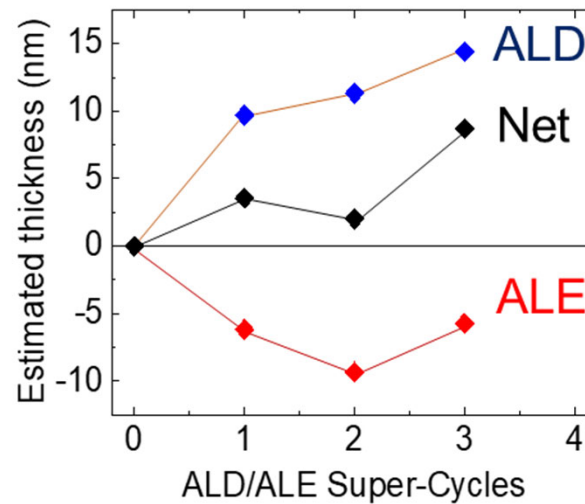
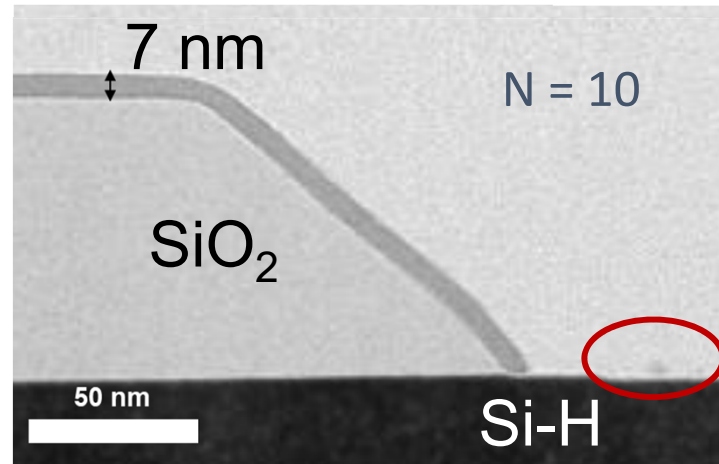
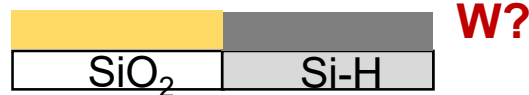
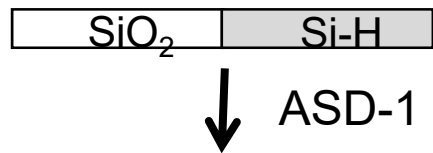


ALD, ALE Rates Inconsistent



Wenyi Xie & Justin Kim (unpublished) 24

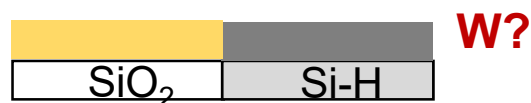
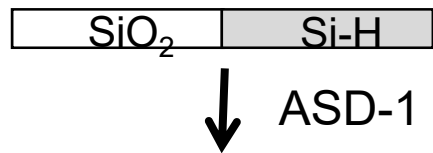
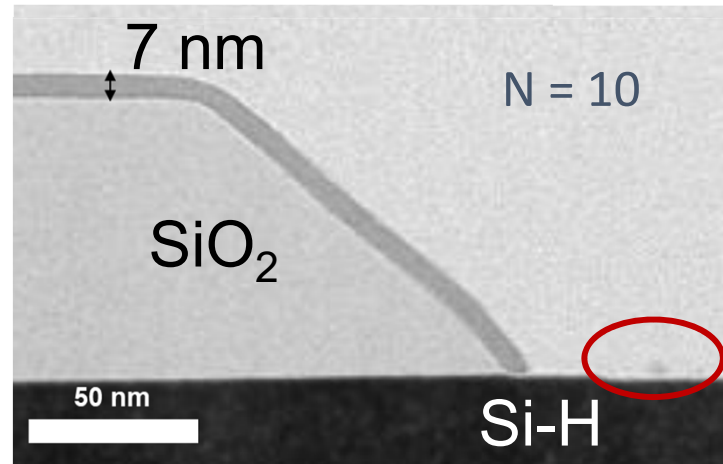
“Dielectric-First” Approach – 2nd try



W ASD vis
ALD+ ALE
doesn't work

Failure

“Dielectric-First” Approach – 3rd try

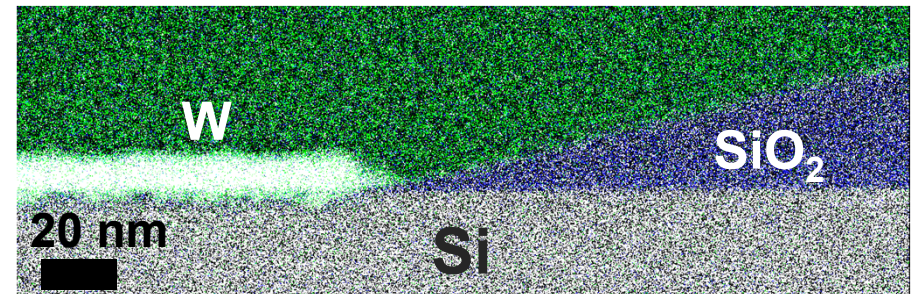
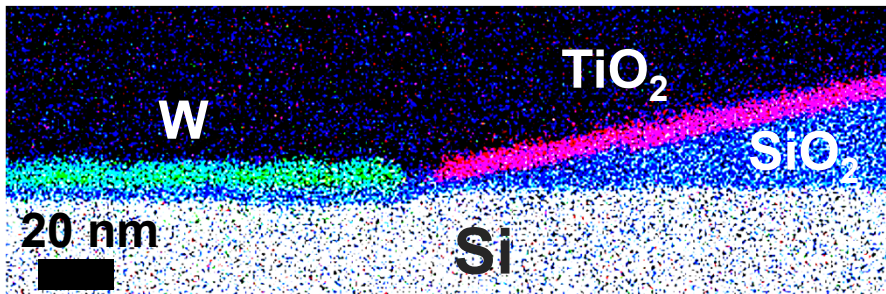
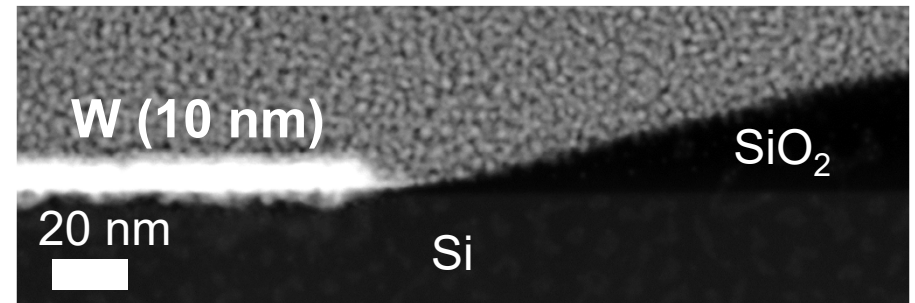
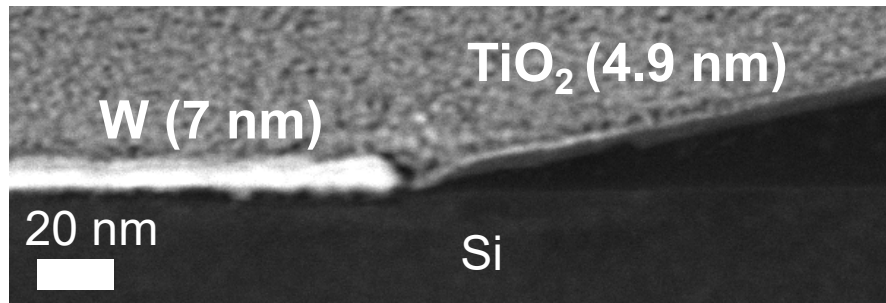


Last chance:

- Do W ALD on TiO₂/Si-H and check nucleation/growth

“Dielectric-First” Approach – 3rd try

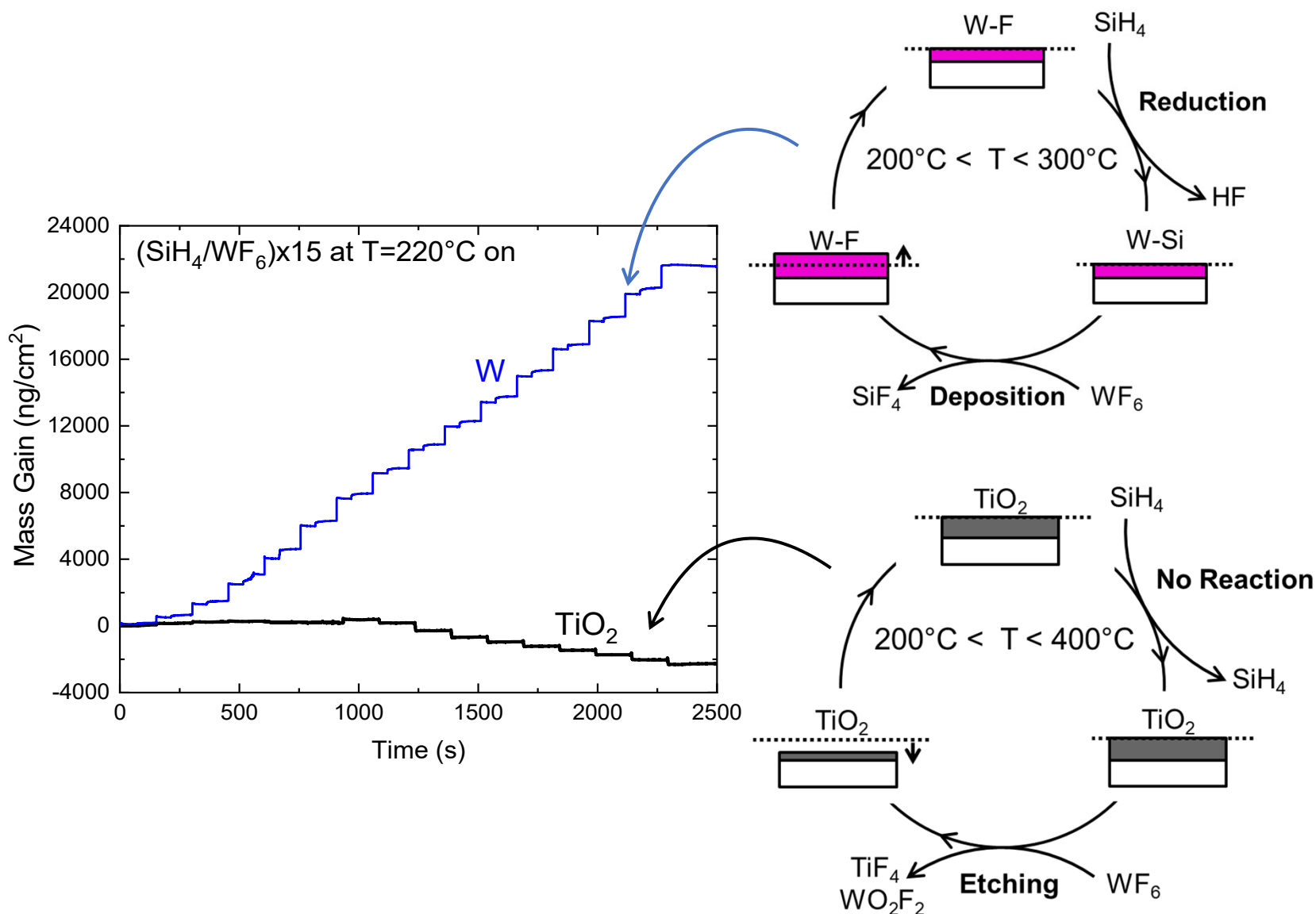
TiO₂ ASD @150°C + W ASD @220°C



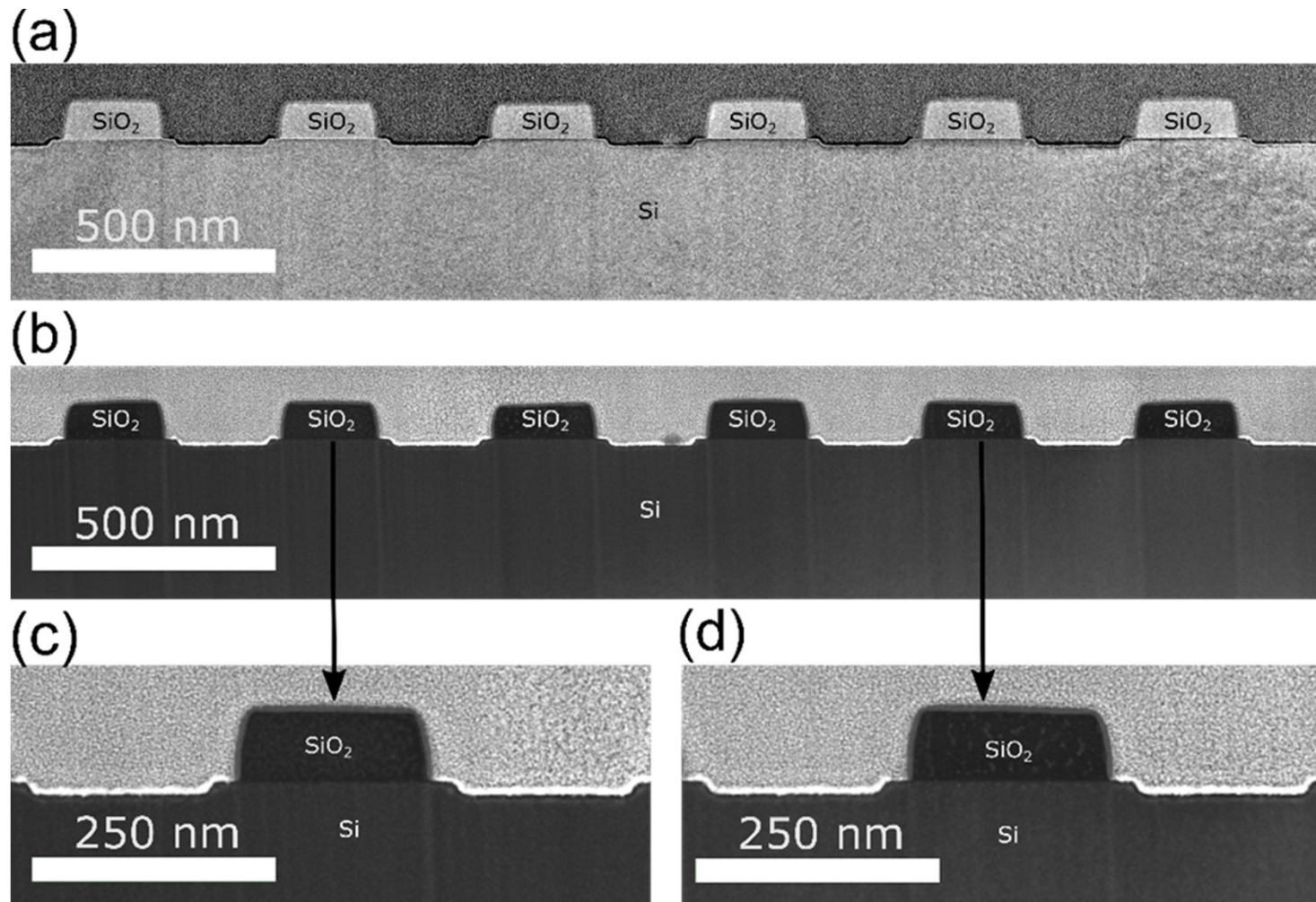
Looks Good!

What is happening?

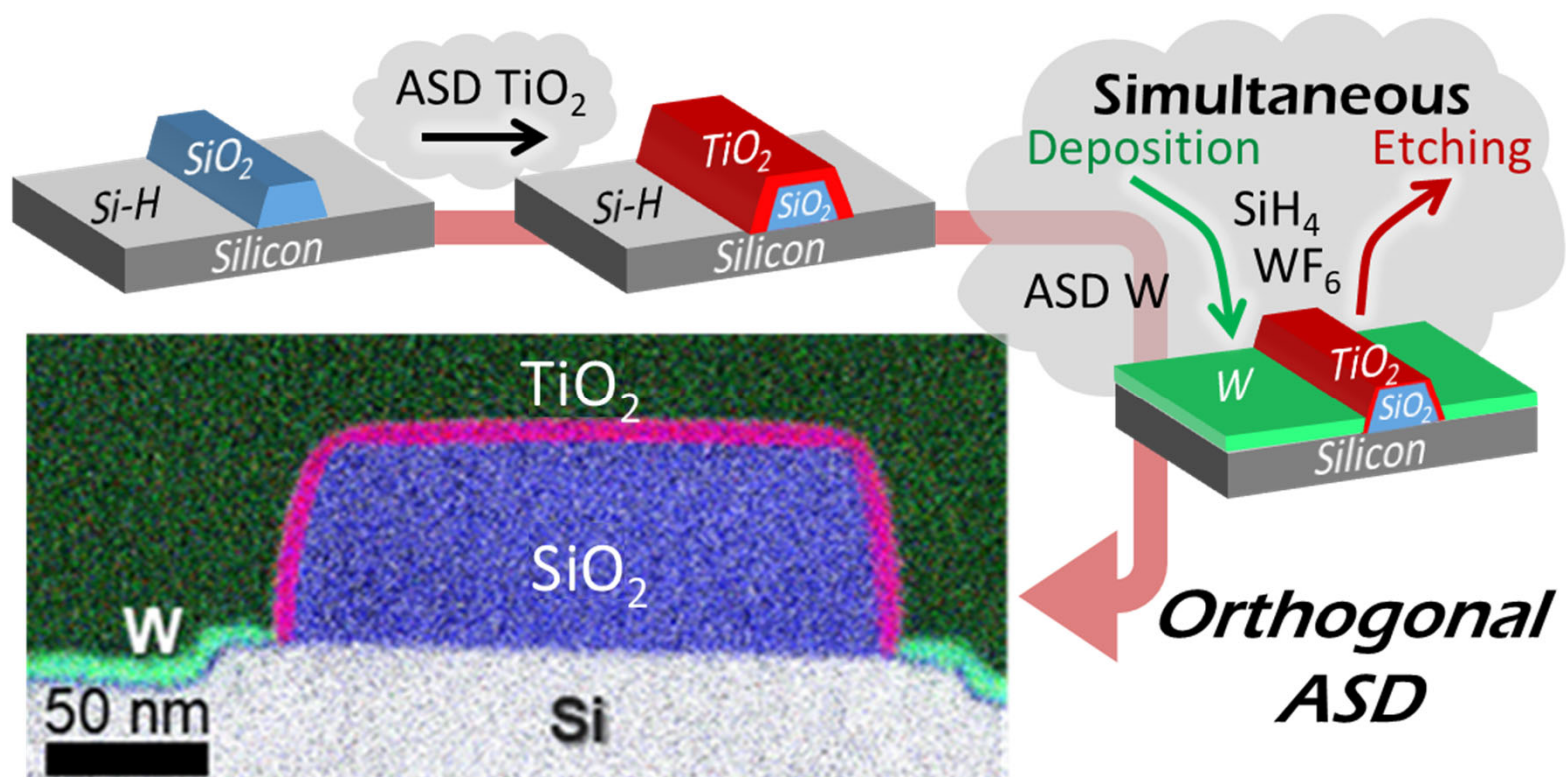
ASD *via* Simultaneous Adjacent Deposition and Etching



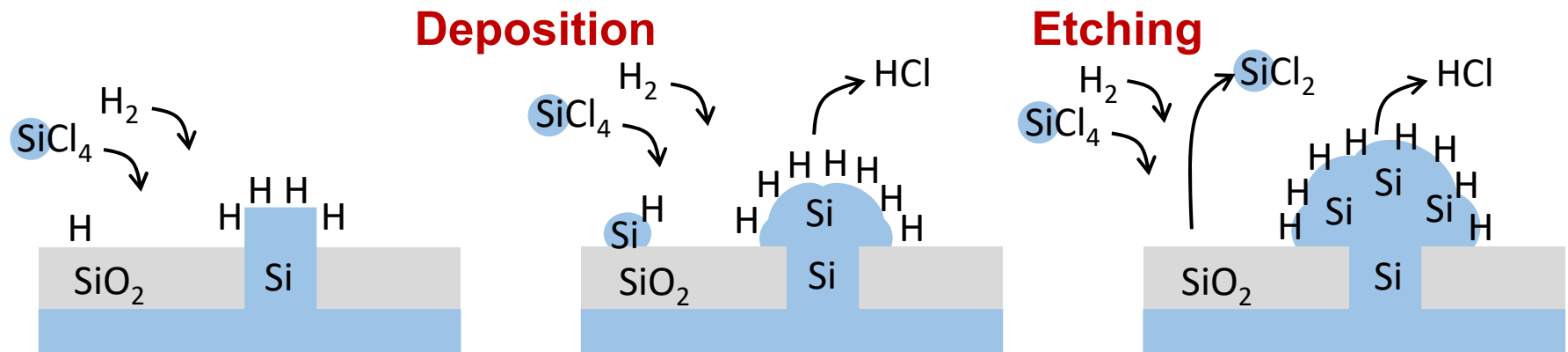
ASD *via* Simultaneous Adjacent Deposition and Etching



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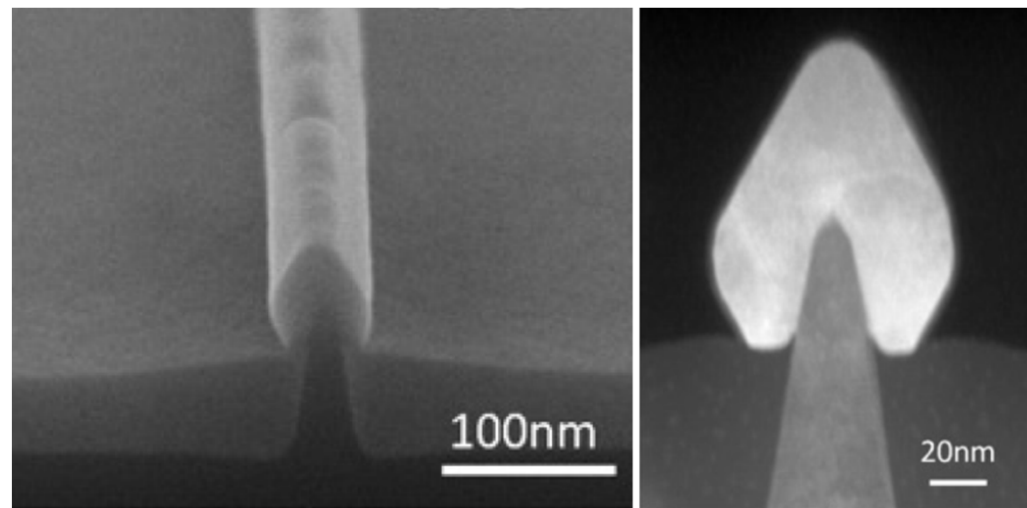


Selective Si, SiGe Epitaxy: High Temperature, $> 700^{\circ}\text{C}$



- ASD controlled by reaction thermodynamics
- Deposition precursor is also an etch reactant
- Analogous to our process, $T < 300^{\circ}\text{C}$

G. Wang et al., *Microelectron. Eng.* (2016)



Alternate Possible Low T ASD Materials

SRC Task 3036.001

Table 2. Partial List of Example Material Systems Showing Thermodynamic Feasibility for Self-Aligned Patterning *via* Simultaneous Deposition and Etching⁴¹

ASD material	Sacrificial material	Reactants	T range (°C)	Expected reactions	Example T (°C)	Rate (nm/cycle)	ΔG (kJ/mol) @ example T	Reference	
W	TiO ₂	SiH ₄ , WF ₆	200 - 300	Deposition	SiH ₄ + WF ₆ → W _(s) + SiF _{4(g)} + HF _(g) + H _{2(g)}	220	0.58	-598	this work, 27
				Etching	TiO ₂ + SiH ₄ + WF ₆ → TiF _{4(g)} + WO ₂ F _{2(g)} + SiH _{4(g)}		0.02 - 0.2	-38	this work, 29
WO ₃	TiO ₂	SiH ₄ , WF ₆ , H ₂ O	200 - 275	Deposition	WF _{6(g)} + SiH _{4(g)} + 3H ₂ O _(g) → WO _{3(s)} + SiF _{4(g)} + 2HF _(g) + 4H _{2(g)}	220	0.58	-653	27, 36
				Etching	TiO _{2(s)} + SiH _{4(g)} + WF _{6(g)} + H ₂ O _(g) → TiF _{4(g)} + WO ₂ F _{2(g)} + SiH _{4(g)} + H ₂ O _(g)		0.02 - 0.2	-38	this work, 29
WO ₃	TiN	H ₂ O, SiH ₄ , WF ₆	200 - 275	Deposition	WF _{6(g)} + SiH _{4(g)} + 3H ₂ O _(g) → WO _{3(s)} + SiF _{4(g)} + 2HF _(g) + 4H _{2(g)}	220	0.58	-653	27, 36
				Etching	TiN _(s) + 2H ₂ O _(g) + SiH _{4(g)} + WF _{6(g)} → TiF _{4(g)} + WO ₂ F _{2(g)} + SiH _{4(g)} + NH _{3(g)} + 0.5H _{2(g)}		0.02 - 0.2	-158	this work, 29
WS ₂	TiO ₂	WF ₆ , H ₂ S	250 - 300	Deposition	H ₂ S + WF ₆ → WS _{2(s)} + HF _(g) + S _{2(g)}	300	~0.1	-409	37
				Etching	TiO ₂ + WF ₆ + H ₂ S → TiF _{4(g)} + WO ₂ F _{2(g)} + H ₂ S _(g)		~0.3	-57	29
AlF ₃	ZnO	HF, TMA	200 - 250	Deposition	Al(CH ₃) _{3(g)} + 3HF _(g) → AlF _{3(s)} + 3CH _{4(g)}	220	~0.05	-701	34
				Etching	ZnO _(s) + 2HF _(g) + 2Al(CH ₃) _{3(g)} → Zn(CH ₃) _{2(g)} + 2AlF(CH ₃) _{2(g)} + H ₂ O _(g)		0.11	< 0 ^b	35
AlF ₃	Ga ₂ O ₃	HF, TMA	200 - 250	Deposition	Al(CH ₃) _{3(g)} + 3HF _(g) → AlF _{3(s)} + 3CH _{4(g)}	250	~0.05	-699	34
				Etching	Ga ₂ O _{3(s)} + 6HF _(g) + 4Al(CH ₃) _{3(g)} → 3H ₂ O _(g) + 2GaF(CH ₃) _{2(g)} + 4AlF(CH ₃) _{2(g)}		~0.082	< 0 ^b	38
Si ₃ N ₄	HfO ₂	HF, SiCl ₄ , NH ₃	300 - 350	Deposition	3SiCl _{4(g)} + 4NH _{3(g)} → Si ₃ N _{4(s)} + 12HCl _(g)	350	0.2	-104	39
				Etching	HfO _{2(s)} + 4HF _(g) + SiCl _{4(g)} → 2H ₂ O _(g) + SiF _{4(g)} + HfCl _{4(g)}		0.005	-77	40
SnS	HfO ₂	HF, Sn(acac) ₂ , H ₂ S	200 - 250	Deposition	Sn(acac) _{2(g)} + H ₂ S _(g) → SnS _(s) + 2H(acac) _(g)	200	0.019	< 0 ^b	41
				Etching	HfO _{2(s)} + 4HF _(g) + 4Sn(acac) _{2(g)} → Hf(acac) _{4(g)} + 4SnF(acac) _(g) + 2H ₂ O _(g)		0.006	< 0 ^b	40
SnS	Al ₂ O ₃	HF, Sn(acac) ₂ , H ₂ S	200 - 250	Deposition	Sn(acac) _{2(g)} + H ₂ S _(g) → SnS _(s) + 2H(acac) _(g)	200	0.019	< 0 ^b	41
				Etching	Al ₂ O _{3(s)} + HF _(g) + Sn(acac) _{2(g)} → Al(acac) _{3(g)} + SnF(acac) _(g) + H ₂ O _(g)		0.023	< 0 ^b	40

Summary

- Motivation: Deposition-based patterning
- Original idea: Orthogonal ASD - Two ASD processes “back-to-back”
- Why we thought our idea would work
 - Success criteria
- Why our original idea was not successful
 - Would the direction have been successful with different boundary conditions?
- How we worked around the problem
 - Discovered a new direction
- What we learned...
 - Success criteria

Success Criteria

- ✓ 1. Create new knowledge about surface reaction processes important for ASD
 - *Deposition and etching can proceed simultaneously at low temperature on adjacent regions of a patterned substrate using a single set of reactants.*
- ✓ 2. Achieve orthogonal ASD of two materials side-by-side (W and TiO₂) with thickness of 5-10 nm each
3. Build up multiple orthogonal layers, ideally 3 or more layers of each ASD material.

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